

FLINN SCIENTIFIC, INC.

Material Safety Data Sheet (MSDS)

MSDS #: 729.00

Revision Date: December 7, 2011

SECTION 1 — CHEMICAL PRODUCT AND COMPANY IDENTIFICATION

Sodium Dichromate Dihydrate

Flinn Scientific, Inc. P.O. Box 219 Batavia, IL 60510 (800) 452-1261

CHEMTREC Emergency Phone Number: (800) 424-9300

SECTION 2 — COMPOSITION, INFORMATION ON INGREDIENTS

Sodium dichromate dihydrate

Synonym: Sodium bichromate; Dichromic acid, disodium salt, dihydrate

CAS#: 7789-12-0

SECTION 3 — HAZARDS IDENTIFICATION

Red or red-orange crystals; deliquescent. Odorless.

Known human carcinogen as a dust. Highly toxic by ingestion, inhalation, and skin absorption.

Corrosive to body tissue. Avoid all body contact.

FLINN AT-A-GLANCE

Health-3

Flammability-0

Reactivity-3

Exposure-3

Storage-3

0 is low hazard, 3 is high hazard

SECTION 4 — FIRST AID MEASURES

Call a physician and seek medical attention for further treatment, observation, and support after first aid.

Inhalation: Remove to fresh air at once. If breathing has stopped, give artificial respiration immediately.

Eye or External: Immediately flush with fresh water for at least 15 minutes.

Internal: Rinse mouth. Give large quantities of water for dilution. Call a physician or poison control at once.

SECTION 5 — FIRE FIGHTING MEASURES

Nonflammable, noncombustible solid.

Strong oxidizer. Fire and explosion risk when in contact with combustible materials. When heated to decomposition, emits toxic fumes of Na₂O.

Fire Fighting Instructions: Use triclass, dry chemical fire extinguisher. Firefighters should wear PPE and SCBA with full facepiece operated in positive mode.

NFPA CODE

None
established

SECTION 6 — ACCIDENTAL RELEASE MEASURES

Restrict unprotected personnel from area. Sweep up, place in sealed bag or container and dispose. Ventilate area and wash spill site after material pickup is complete. See Sections 8 and 13 for further information.

SECTION 7 — HANDLING AND STORAGE

Flinn Suggested Chemical Storage Pattern: Inorganic #8. Store with borates, chromates, manganates and permanganates.

Store in a cool dry place. Keep container tightly closed. Store in a Flinn Chem-Saf bag, and lock up in a poison cabinet.

Use and dispense in a hood.

SECTION 8 — EXPOSURE CONTROLS, PERSONAL PROTECTION

Avoid contact with eyes, skin, and clothing. Wear chemical splash goggles, chemical-resistant gloves, and chemical-resistant apron. Use ventilation to keep airborne concentrations below exposure limits. Always wear a NIOSH-approved respirator with proper cartridges or a positive pressure, air-supplied respirator when handling this material in emergency situations (spill or fire).

Exposure guidelines: PEL 0.05 mg/m³ (as Cr), Ceiling: 0.1 mg/m³ (as CrO₃) (ACGIH) (OSHA)

SECTION 9 — PHYSICAL AND CHEMICAL PROPERTIES

Red or red-orange crystals; deliquescent. Odorless.

Solubility: Soluble in water; not alcohol.

Formula: Na₂Cr₂O₇ · 2H₂O

Formula weight: 297.999

Melting point: 85 °C (decomposes)

Specific gravity: 2.35

SECTION 10 — STABILITY AND REACTIVITY

Avoid contact with strong reducers, and alcohols.

Shelf life: Fair to poor, deliquescent. Keep tightly closed.

SECTION 11 — TOXICOLOGICAL INFORMATION

Acute effects: Poison, corrosive, sensitizer

Chronic effects: Carcinogen, mutagen

Target organs: N.A.

ORL-RAT LD₅₀: 52 mg/kg

IHL-RAT LC₅₀: 124 mg/m³

SKN-RBT LD₅₀: 1.00 g/kg

N.A. = Not available, not all health aspects of this substance have been fully investigated.

SECTION 12 — ECOLOGICAL INFORMATION

Data not yet available.

SECTION 13 — DISPOSAL CONSIDERATIONS

Please review all federal, state and local regulations that may apply, before proceeding.

Flinn Suggested Disposal Method #12a is one option.

SECTION 14 — TRANSPORT INFORMATION

Shipping name: Oxidizing Solid, n.o.s.

Hazard class: 5.1, Oxidizer

UN number: N/A

N/A = Not applicable

SECTION 15 — REGULATORY INFORMATION

TSCA-listed, EINECS-listed (234-190-3), RCRA code D001, D007.

SECTION 16 — OTHER INFORMATION

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**Consult your copy of the *Flinn Science Catalog/Reference Manual*
for additional information about laboratory chemicals.**